

Preliminary Amendment  
Application No.: 10/820,024  
Reply to Office Action dated September 25, 2006  
March 26, 2007

AMENDMENTS TO THE CLAIMS

Please substitute the following claims for the pending claims with the same numbers respectively:

Claims 1-2 (Cancelled):

Claim 3 (Currently amended): A method for making a conductive electroless plating powder comprising the steps of:

(I) allowing core particles which have a noble metal ion-capturing ability to capture noble metal ions, and reducing the noble metal ions so that the surfaces of the core particles support the noble metal;

(II) dispersing the core particles in an initial thin film-forming solution containing nickel ions, a reducing agent, and a complexing agent comprising an amine to prepare an aqueous suspension, and reducing the nickel ions to form initial thin nickel films on a surface of each of the core particles; and

(III) adding a first solution, which contains a nickel ion-containing solution containing and the complexing agent, and a second solution, which contains a reducing agent-containing solution individually and simultaneously, to the aqueous suspension individually and simultaneously, the aqueous

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suspension containing the core particles provided with the initial thin nickel films and the complexing agent to perform electroless plating so that columnar structures extending in a direction of a thickness of a nickel film are formed,  
adjusting the amounts of the nickel ion-containing solution added and the reducing agent-containing solution added,  
adjusting the initial concentration of the complexing agent in the aqueous suspension; and  
adjusting the concentration of the complexing agent in the nickel ion-containing solution, so as to maintain the concentration of the complexing agent in the aqueous suspension in the range of 0.003 to 10 moles/l in said step of (III) adding a nickel ion-containing solution containing the same complexing agent and a reducing agent-containing solution.

Claim 4 (Cancelled):

Claim 5 (Previously presented): The method according to claim 3, further comprising the step of using glycine or ethylenediamine as the complexing agent.

Claim 6 (Cancelled):

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Claim 7 (Previously presented): The method according to claim 3, further comprising the step of providing, before said step (III), a ratio of the sum of the surface areas of the core particles contained in the aqueous suspension to the volume of the aqueous suspension between 0.1 to 15 m<sup>2</sup>/l.

Claim 8 (Cancelled):

Claim 9 (Previously presented): The method according to claim 5, further comprising the step of providing, before said step (III), a ratio of the sum of the surface areas of the core particles contained in the aqueous suspension to the volume of the aqueous suspension between 0.1 to 15 m<sup>2</sup>/l.

Claim 10 (Previously presented): The method according to claim 3, further comprising the step of imparting the noble metal ion-capturing ability to the core particles by a surface treatment.

Claim 11 (Previously presented): The method according to claim 10, wherein said step of imparting the noble metal ion-capturing ability to the core particles by a surface treatment includes adjusting the amount of the surface treatment in the

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range between 0.3 and 100 mg/m<sup>2</sup> of the surface area of the core particles.

Claim 12 (Currently amended): The method according to claim 3, wherein said step of (II) dispersing the core particles in an initial thin film-forming solution containing nickel ions, a reducing agent, and a complexing agent includes using glycine or ethylenediamine for the complexing agent; and

wherein said step of (III) adding a first solution, which contains a nickel ion-containing solution ~~containing~~ and the complexing agent, and a second solution, which contains a reducing agent-containing solution, includes using glycine or ethylenediamine for the complexing agent.

Claim 13 (Previously presented): The method according to claim 3, wherein said step of adjusting the initial concentration of the complexing agent includes using glycine or ethylenediamine.

Claim 14 (Currently amended): The method according to claim 3, further comprising the step of providing, before said step of (III) adding a first solution, which contains a nickel ion-containing solution ~~containing~~ and the complexing agent, and a

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second solution, which contains a reducing agent-containing solution, a ratio of the sum of the surface areas of the core particles contained in the aqueous suspension to the volume of the aqueous suspension between 0.1 to 15 m<sup>2</sup>/l.

Claim 15 (Currently amended): The method according to claim 12, further comprising the step of providing, before said step of (III) adding a first solution, which contains a nickel ion-containing solution containing and the complexing agent, and a second solution, which contains a reducing agent-containing solution, a ratio of the sum of the surface areas of the core particles contained in the aqueous suspension to the volume of the aqueous suspension between 0.1 to 15 m<sup>2</sup>/l.

Claim 16 (Currently amended): A method for making a conductive electroless plating plated powder including columnar structures extending in a direction of a thickness of a nickel film comprising the steps of:

(I) allowing the core particles which have a noble metal ion-capturing ability to capture noble metal ions, and reducing the noble metal ions so that the surfaces of the core particles support the noble metal;

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(II) dispersing the core particles in an initial thin film-forming solution containing nickel ions, a reducing agent, and a complexing agent comprising an amine to prepare an aqueous suspension, and reducing the nickel ions to form initial thin nickel films on a surface of each of the core particles;

~~providing a ratio of the sum of the surface areas of the core particles contained in the aqueous suspension to the volume of the aqueous suspension between 0.1 to 15 m<sup>2</sup>/l;~~

(III) adding a first solution, which contains a nickel ion-containing solution containing and the [[same]] complexing agent, and a second solution, which contains a reducing agent-containing solution individually and simultaneously, to the aqueous suspension individually and simultaneously, the aqueous suspension containing the core particles provided with the initial thin nickel films and the complexing agent to perform electroless plating;

~~adjusting the amounts of the nickel ion-containing solution added and the reducing agent-containing solution added;~~

~~adjusting the initial concentration of the complexing agent in the aqueous suspension;~~

~~adjusting the concentration of the complexing agent in the nickel ion-containing solution, so as to maintain the concentration of the complexing agent in the aqueous suspension~~

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~~in the range of 0.003 to 10 moles/l in said step of adding a nickel ion-containing solution containing the same complexing agent and a reducing agent-containing solution.~~

Claim 17 (Currently amended): The method according to claim 16, wherein said step of allowing core particles includes dispersing the core particles in a ~~weakly~~ weakly acidic aqueous solution of a noble metal salt which is palladium chloride so that the noble metal ions are captured by the surfaces of the core particles.

Claim 18 (Previously presented): The method according to claim 16, wherein said step of dispersing the core particles in an initial thin film-forming solution containing nickel ions includes forming an initial thin film in the range of the thickness between 0.001 and 2  $\mu\text{m}$ .

Claim 19 (Previously presented): The method according to claim 18, wherein said step of dispersing the core particles in an initial thin film-forming solution containing nickel ions includes forming an initial thin film in the range of the thickness between 0.005 and 1  $\mu\text{m}$ .

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Claim 20 (Previously presented): The method according to claim 16, wherein said step of dispersing the core particles in an initial thin film-forming solution containing nickel ions includes adjusting the concentration of the nickel ions in the initial thin film-forming solution in the range between  $2.0 \times 10^{-4}$  and 1.0 mol/l.

Claim 21 (Previously presented): The method according to claim 20, wherein said step of dispersing the core particles in an initial thin film-forming solution containing nickel ions includes adjusting the concentration of the nickel ions in the initial thin film-forming solution in the range between  $1.0 \times 10^{-3}$  and 0.1 mol/l.

Claim 22 (Previously presented): The method according to claim 16, wherein said step of dispersing the core particles in an initial thin film-forming solution containing nickel ions includes adjusting the reducing agent in the initial thin film-forming solution in the range between  $4 \times 10^{-4}$  and 2.0 mol/l.

Claim 23 (Previously presented): The method according to claim 22, wherein said step of dispersing the core particles in an initial thin film-forming solution containing nickel ions

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includes adjusting the reducing agent in the initial thin film-forming solution in the range between  $2.0 \times 10^{-3}$  and 0.2 mol/l.

Claim 24 (Previously presented): The method according to claim 16, wherein said step of dispersing the core particles in an initial thin film-forming solution containing nickel ions, a reducing agent, and a complexing agent includes using sodium borohydride or dimethylamine borane for the reducing agent.

Claim 25 (Previously presented): The method according to claim 16, wherein said step of dispersing the core particles in an initial thin film-forming solution containing a complexing agent includes using glycine for the complexing agent; and

wherein said step of adding a nickel ion-containing solution containing the complexing agent includes using glycine for the complexing agent.

Claim 26 (Currently amended): The method according to claim [[16]] 37, wherein said step of adjusting the initial concentration of the complexing agent includes using glycine.

Claim 27 (Currently amended): The method according to claim 16, wherein said step of (III) adding a first solution, which

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contains a nickel ion-containing solution containing and the complexing agent, and a second solution, which contains a reducing agent-containing solution, includes adjusting the concentration of the nickel ions in the nickel ion-containing solution in the range between 0.1 and 1.2 mol/l.

Claim 28 (Currently amended): The method according to claim 27, wherein said step of (III) adding a first solution, which contains a nickel ion-containing solution containing and the complexing agent, and a second solution, which contains a reducing agent-containing solution, includes adjusting the concentration of the nickel ions in the nickel ion-containing solution in the range between 0.5 and 1.0 mol/l.

Claim 29 (Currently amended): The method according to claim 16, wherein said step of (III) adding a first solution, which contains a nickel ion-containing solution containing and the complexing agent, and a second solution, which contains a reducing agent-containing solution, includes adjusting the concentration of the reducing agent in the nickel ion-containing solution in the range between 0.1 and 20 mol/l.

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Claim 30 (Currently amended): The method according to claim 29, wherein said step of (III) adding a nickel ion-containing solution containing and the complexing agent, and a second solution, which contains a reducing agent-containing solution, includes adjusting the concentration of the reducing agent in the nickel ion-containing solution in the range between 1 and 10 mol/l.

Claim 31 (Currently amended): The method according to claim 16, wherein said step of (III) adding a first solution, which contains a nickel ion-containing solution containing and the complexing agent, and a second solution, which contains a reducing agent-containing solution, includes adjusting the deposition rate of nickel between 1 and 10,000 nanometers/hour.

Claim 32 (Currently amended): The method according to claim 31, wherein said step of (III) adding a first solution, which contains a nickel ion-containing solution containing and the complexing agent, and a second solution, which contains a reducing agent-containing solution, includes adjusting the deposition rate of nickel between 5 and 300 nanometers/hour.

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Claim 33 (Currently amended): A method for making a conductive electroless plating plated powder comprising the steps of:

(I) allowing the core particles which have a noble metal ion-capturing ability to capture noble metal ions, and reducing the noble metal ions so that the surfaces of the core particles support the noble metal;

(II) dispersing the core particles in an initial thin film-forming solution containing nickel ions, a reducing agent, and a complexing agent comprising an amine to prepare an aqueous suspension, and reducing the nickel ions to form initial thin nickel films on a surface of each of the core particles;

(III) adding a first solution, which contains a nickel ion-containing solution containing and the [[same]] complexing agent, and a second solution, which contains a reducing agent-containing solution individually and simultaneously to the aqueous suspension individually and simultaneously, the aqueous suspension containing the core particles provided with the initial thin nickel films and the complexing agent to perform electroless plating so that columnar structures extending in a direction of a thickness of a nickel film are formed; and

(IV) forming a gold plating plated layer as a top layer on the nickel film.

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Claim 34 (Currently amended): A method for making a conductive electroless ~~plating~~ plated powder including columnar structures extending in a direction of a thickness of a nickel film comprising the steps of:

(I) allowing the core particles which have a noble metal ion-capturing ability to capture noble metal ions, and reducing the noble metal ions so that the surfaces of the core particles support the noble metal;

(II) dispersing the core particles in an initial thin film-forming solution containing nickel ions, a reducing agent, and a complexing agent comprising an amine to prepare an aqueous suspension, and reducing the nickel ions to form initial thin nickel films on a surface of each of the core particles;

~~providing a ratio of the sum of the surface areas of the core particles contained in the aqueous suspension to the volume of the aqueous suspension between 0.1 to 15 m<sup>2</sup>/l;~~

(II) adding a first solution, which contains a nickel ion-containing solution containing and the [[same]] complexing agent, and a second solution, which contains a reducing agent-containing solution individually and simultaneously, to the aqueous suspension individually and simultaneously, the aqueous suspension containing the core particles provided with the

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initial thin nickel films and the complexing agent to perform electroless plating;

~~adjusting the amounts of the nickel ion-containing solution added and the reducing agent-containing solution added;~~

~~adjusting the initial concentration of the complexing agent in the aqueous suspension;~~

~~adjusting the concentration of the complexing agent in the nickel ion-containing solution, so as to maintain the concentration of the complexing agent in the aqueous suspension in the range of 0.003 to 10 moles/l in said step of adding a nickel ion-containing solution containing the same complexing agent and a reducing agent-containing solution; and~~

(IV) forming a gold plating plated layer as a top layer on the nickel film.

Please add the following new claims 35-39 as follows:

Claim 35 (New): The method according to claim 3, further comprising at least one of the steps of:

~~adjusting the amounts of the nickel ion-containing solution added and the reducing agent-containing solution added;~~

~~adjusting the initial concentration of the complexing agent in the aqueous suspension; and~~

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adjusting the concentration of the complexing agent in the nickel ion-containing solution, so as to maintain the concentration of the complexing agent in the aqueous suspension in the range of 0.003 to 10 moles/l in said step of (III) adding the first and second solutions.

Claim 36 (New): The method according to claim 16, further comprising the step of providing a ratio of the sum of the surface areas of the core particles contained in the aqueous suspension to the volume of the aqueous suspension between 0.1 to 15 m<sup>2</sup>/l.

Claim 37 (New): The method according to claim 16, further comprising at least one of the steps of:

adjusting the amounts of the nickel ion-containing solution added and the reducing agent-containing solution added;

adjusting the initial concentration of the complexing agent in the aqueous suspension; and

adjusting the concentration of the complexing agent in the nickel ion-containing solution, so as to maintain the concentration of the complexing agent in the aqueous suspension in the range of 0.003 to 10 moles/l in said step of adding the first and second solutions.

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Claim 38 (New): The method according to claim 34, further comprising the step of providing a ratio of the sum of the surface areas of the core particles contained in the aqueous suspension to the volume of the aqueous suspension between 0.1 to 15 m<sup>2</sup>/l.

Claim 39 (New): The method according to claim 34, further comprising at least one of the steps of:

adjusting the amounts of the nickel ion-containing solution added and the reducing agent-containing solution added;

adjusting the initial concentration of the complexing agent in the aqueous suspension; and

adjusting the concentration of the complexing agent in the nickel ion-containing solution, so as to maintain the concentration of the complexing agent in the aqueous suspension in the range of 0.003 to 10 moles/l in said step of adding a first solution, which contains a nickel ion-containing solution and the complexing agent, and a second solution, which contains a reducing agent-containing solution.